# metal-organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.007 Å R factor = 0.048 wR factor = 0.099 Data-to-parameter ratio = 15.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# cis-Dibromobis(1,10-phenanthroline)manganese(II)

In the title complex,  $[MnBr_2(C_{24}H_{16}N_4)]$ , there is a weak C-H···Br intermolecular interaction in the crystal structure, giving rise to molecular chains along the *a* axis.

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#### Comment

In recent years, simple metal complexes of phenanthroline and its derivatives have attracted great interest because they can be used to study the hydrolysis of biologically important phosphate diesters with poor leaving groups, *e.g.* DNA (Wall *et al.*, 1999). These complexes can also be used to develop new diagnostic and therapeutic agents in DNA binding and cleavage (Barton, 1986; Naing *et al.*, 1995). We report here the title complex, (I), a new metal complex of phenanthroline. A search of the January 2004 Cambridge Structural Database (Allen, 2002; Bruno *et al.*, 2002) found no report of (I).



Compound (I) has expected values for bond lengths and angles. The dihedral angle between the two phenanthroline planes is 84.74 (6)° and the average Mn—Br bond length is 2.535 (2) Å, within the range of previously reported values [2.496 (2) (Kienitz *et al.*, 2000) and 2.677 (2) Å (Goodgame *et al.*, 1999)]. There is a weak intermolecular interaction, C13—



© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved The molecular structure of (I), shown with 30% probability displacement ellipsoids.



**Figure 2** The packing of (I), viewed down the *a* axis.



#### Figure 3

The molecular chains formed by the weak intermolecular interaction (dashed lines) of C13-H13···Br2<sup>i</sup> [symmetry code: (i)  $x, \frac{3}{2} - y, \frac{1}{2} + z.$ ]

H13···Br2<sup>i</sup> [symmetry code: (i)  $x, \frac{3}{2} - y, \frac{1}{2} + z$ ], in the structure, giving one-dimensional molecular chains arranged along the *a* axis, as shown in Figs. 2 and 3.

## **Experimental**

To a warm solution of 1,10-phenanthroline (0.344 g, 2 mmol) in CH<sub>3</sub>OH (20 ml) was added MnBr<sub>2</sub>·4H<sub>2</sub>O (0.265 g, 1 mmol) with slow heating and stirring. The mixture was refluxed for 2 h, then cooled to room temperature. After about two weeks, yellow single crystals suitable for X-ray crystallographic analysis were obtained. Analysis calculated for C<sub>24</sub>H<sub>16</sub>Br<sub>2</sub>MnN<sub>4</sub>: C 50.11, H 2.80, N 9.74%; found: C 50.05, H 2.76, N 9.69%.

#### Crystal data

$[MnBr_2(C_{24}H_{16}N_4)]$	$D_x = 1.730 \text{ Mg m}^{-3}$
$M_r = 575.17$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 1225
a = 10.423 (3) Å	reflections
b = 16.223 (3) Å	$ heta=2.5 extrm{-}20.4^\circ$
c = 13.244 (3) Å	$\mu = 4.24 \text{ mm}^{-1}$
$\beta = 99.62 \ (2)^{\circ}$	T = 293 (2)  K
V = 2208.0 (9) Å <sup>3</sup>	Block, yellow
Z = 4	$0.20 \times 0.10 \times 0.10$ mm

#### Data collection

Bruker SMART APEX CCD area- detector diffractometer $\varphi$ and $\omega$ scans	4334 independent reflections 3284 reflections with $I > 2\sigma(I)$ $R_{int} = 0.042$
Absorption correction: multi-scan	$\theta_{\rm max} = 26.0^{\circ}$
(SADABS; Bruker, 2000)	$h = -12 \rightarrow 12$
$T_{\min} = 0.61, T_{\max} = 0.66$	$k = -20 \rightarrow 20$
9269 measured reflections	$l = -16 \rightarrow 9$
Refinement	
Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	+ 1.55P]
$wR(F^2) = 0.099$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} < 0.001$
4334 reflections	$\Delta \rho_{\rm max} = 0.96 \ {\rm e} \ {\rm \AA}^{-3}$
280 parameters	$\Delta \rho_{\rm min} = -0.73 \mathrm{e} \mathrm{\AA}^{-3}$
H-atom parameters constrained	

### Table 1

Selected geometric parameters (Å, °).

Br1_Mn1	2 5313 (10)	Mn1_N3	2429(4)
Br2-Mn1	2.5268(11)	Mn1-N2	2.425(4) 2.445(4)
Mn1-N4	2.391 (4)	Mn1-N1	2.448 (4)
N4 N4 1 N2	(7.0( (12)		00.06 (0)
N4-Mn1-N3	67.06 (13)	N2-Mn1-Br2	99.86 (9)
N4-Mn1-N2	95.51 (13)	N1-Mn1-Br2	160.63 (9)
N3-Mn1-N2	161.56 (12)	N4-Mn1-Br1	161.29 (11)
N4-Mn1-N1	80.27 (13)	N3-Mn1-Br1	99.77 (9)
N3-Mn1-N1	107.97 (12)	N2-Mn1-Br1	95.62 (10)
N2-Mn1-N1	61.13 (12)	N1-Mn1-Br1	92.04 (9)
N4-Mn1-Br2	99.19 (11)	Br2-Mn1-Br1	93.66 (3)
N3-Mn1-Br2	89.27 (9)		

#### **Table 2** Hydrogen-bonding geometry (Å, °).

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$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$		
$C13-H13\cdots Br2^{i}$	0.93	2.78	3.466 (5)	131		
6	1.					

Symmetry code: (i)  $x, \frac{3}{2} - y, \frac{1}{2} + z$ .

All H atoms were positioned geometrically (C-H = 0.93 Å) and treated as riding, with  $U_{iso}(H) = 1.2U_{eq}(C)$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve and refine structure: *SHELXTL* (Bruker, 2000); molecular graphics: *SHELXTL* and *MERCURY* (Version 1.2.1; Bruno *et al.*, 2002).

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